### Form Approved REPORT DOCUMENTATION PAGE OMB No. 0704-0188 Public reporting burden for this collection of information is estimated to average 1 hour per response, including the time for reviewing instructions, searching data sources, gathering and maintaining the data needed, and completing and reviewing the collection of information. Send comments regarding this burden estimate or any other aspect of this collection of information, including suggestions for reducing this burden to Washington Headquarters Service, Directorate for Information Operations and Reports, 1215 Jefferson Davis Highway, Suite 1204, Airlighor, VA 22202-4302, and to the Office of Management and Budget, Paperwork Reduction Project (0704-0188) Washington, DC 20503. PLEASE DO NOT RETURN YOUR FORM TO THE ABOVE ADDRESS. 3. DATES COVERED (From - To) 1. REPORT DATE (DD-MM-YYYY) 2. REPORT DATE June 99 - Dec 99 Final 30-01-2003 5a. CONTRACT NUMBER 4. TITLE AND SUBTITLE A New Class of Highly Polar Liquid Crystals 5b. GRANT NUMBER For Display Applications N 000 14-99-1-0854 5c. PROGRAM ELEMENT NUMBER 5d. PROJECT NUMBER 6. AUTHOR(S) Piotr Kaszynski 5e. TASK NUMBER 5f. WORK UNIT NUMBER 7. PERFORMING ORGANIZATION NAME(S) AND ADDRESS(ES) 8. PERFORMING ORGANIZATION REPORT NUMBER Vanderbilt University Nashville, TN 37235 10. SPONSOR/MONITOR'S ACRONYM(S) 9. SPONSORING/MONITORING AGENCY NAME(S) AND ADDRESS(ES) ONR Office of Naval Research 11. SPONSORING/MONITORING Ballston Centre Tower One AGENCY REPORT NUMBER 800 N Quincy Street Arlington, VA 22217-5660 12. DISTRIBUTION AVAILABILITY STATEMENT

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#### 14. ABSTRACT

The subject of this project is the synthesis and characterization of a new class of molecular components for nematic-based liquid crystal displays that show good solubility in the nematic host and high efficiency at low doping levels

The centerpiece of the design is the inorganic boron cluster CB11H12(-) which upon substitution with a positively charged group forms highly polar, UV transparent liquid crystal materials.

### 15. SUBJECT TERMS

liquid crystals, polarity, carboranes, synthesis

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# Final report from work supported by ONR-331/99/0237 grant

# A New Class of Highly Polar Liquid Crystals for Display Applications

#### I. Goals

- Development of new liquid crystalline components for nematic mixtures which will lower the operational voltage of devices, increase photostability, and permit construction of thin film cells.
- The key structural element for the design of this new class of liquid crystals is an inorganic 12-vertex monocarbaborate cluster, CB<sub>11</sub>H<sub>12</sub>(-), whose negative charge is compensated with a positively charged substituent Q (Figure 1).

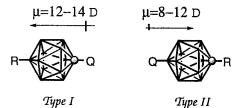


Figure 1. Two types of betaines derived from carbadodecaborate anion CB<sub>11</sub>H<sub>12</sub>(-1). Each unsubstituted vertex corresponds to a BH fragment and the sphere represents a carbon atom. R is an alkyl group and Q is an onium substituent such as pyridinium, quinuclidinium, sulfonium and oxonium. The arrows indicate the direction and the relative magnitudes of dipole moments  $\mu$ .

- Key features of the new materials:
  - 1. Large dipole moments of about 12-14 D rigidly oriented along the long molecular axes. (This property gives a lower operational voltage for the electrooptical device)
  - 2. UV transparency above 210 nm. (This increases photostability of the material.)
  - 3. Low birefringence and high refractive index. (This allows for thinner layers of the liquid crystal material.)

### II Synthetic Strategy

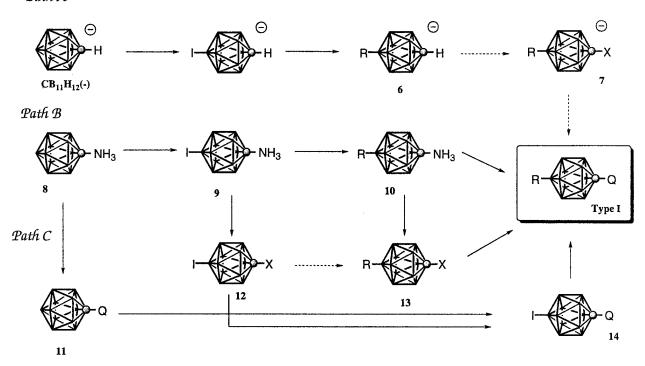
The onium group Q has been chosen as: quinuclidine (1), tetrahydropyran (2, 4) and tetrahydrothiopyran (3, 5), each substituted with pentyl groups. The alkyl group attached to the cage was initially chosen to be either pentyl (a) or decyl (b).

Type 
$$I$$
  $C_5H_{11}$   $C_5H_{1$ 

 $a, R=C_5H_{11}, b, R=C_{10}H_{21},$ 

• Our initial focus was on Type I liquid crystals and Scheme 1 shows synthetic pathways to this class of materials.

Path A



- Path A relies on C-functionalization of the 12-alkyl derivative 6 whose reliable synthesis was recently described in the literature (Gruner et al, J. Am. Chem. Soc., 1999, 121, 3122).
- Path B starts from C-amino derivative 8 which is substituted in the 12 position with an alkyl group. The amino group serves as a synthetic handle to introduce the two other heteroatoms via diazotization either as X (OH<sub>2</sub> and SH<sub>2</sub>) or onium groups Q, and is also converted into the C-quinuclidine derivative 1.
- Path C is derived from the method, which we used previously to prepare the first quantities of the quinuclidine derivatives 1a and 1b.

### III. Challenges

- 1,12 Difunctionalization of the  $CB_{11}H_{12}$  cluster was virtually unknown and the first approaches were developed by us (Path C).
- Characterization of electro-optical performance of highly polar materials.

#### IV. Results

#### **Synthesis**

- The original synthesis of 1 along Path C was very inefficient.
- Significant difficulties were encountered with C-functionalization of 12-alkyl carbaborates. Reactions with electrophiles that typically give moderate to good yields for the parent cluster CB<sub>11</sub>H<sub>12</sub>(-), were ineffective in the case of 12-alkyl derivatives 6 (Path A).
- We discovered that iodination of amine 8 gives good yields of the 12-iodo derivative 9, which, in turn, undergoes particularly efficient Pd-catalyzed coupling reactions to yield 12-alkyl derivatives 10 (Path B).
- Following Path B, we have prepared significant quantities (>0.25 g) of the quinuclidine derivatives 1 and completed their characterization including single crystal X-ray analysis.
- The C-amino derivative 10 was successfully converted into the C-hydroxy derivative 13 (X=OH<sub>2</sub>) by diazotization. The alkylation step turned out to be more difficult than we anticipated but we found solution to this problem.
- A reproducible and efficient synthesis of the **13** (X=SH<sub>2</sub>) from the C-amino derivative **10** still poses a challenge despite the initial observation of the desired product.

#### Characterization

- The dipole moment for 11 (Q = quinuclidine) was measured to be 12 D (calculated 14 D)
- Solubility in a nematic host for 1 has been evaluated
- Electrooptical performance of 1 in two nematic hosts has been done
- Roentgenographic characterization of the LC phases for 1 is complete.

#### Computational analysis

All calculations have been performed using ab initio quantum-mechanical methods.

- Dipole moments have been calculated and compared with the experimental value.
- Conformational analysis for the 1 and 2 is almost complete.
- Evaluation of energy barriers to inversion at the O and S centers (in 2 and 3, respectively) is begun. (This is an important factor governing the isomeric composition for the compounds).
- Account for different reactivity of 12-iodo derivatives towards Grignard reagents is complete.

### V. Accomplishments

- A facile methodology for 1-12 diffunctionalization of the CB<sub>11</sub>H<sub>12</sub> clusters
- Practical synthesis of the quinuclidine derivatives 1
- Identification of the LC phases by powder XRD for 1
- Studies of electrooptical performance of the quinuclidine derivatives 1

• Theoretical understanding molecular properties of the mesogens.

## VI. Publications with acknowledgment of the grant support

- 1. Pakhomov, S.; Kaszynski, P.; Young, V. G. Jr. Inorg. Chem., 2000, 39, 2243-2245.
- 2. Kaszynski, P. in Anisotropic Organic Materials-Approaches to Polar Order, R. Glaser and P. Kaszynski, Eds. ACS Symposium Series, Volume 798. American Chemical Society, Washington, D.C. 2001, pp 68-82.
- 3. Pakhomov, S.; Douglass, A. G.; Balisnki, A.; Kaszynski, P., J. Am. Chem. Soc., in preparation.